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Spin-Lock MR Enhances the Detection Sensitivity of Superparamagnetic Iron Oxide Particles

Rik P. M. Moonen,1 Pieterien van der Tol,1,2 Stefanie J. C. G. Hectors,1 Lucas W. E. Starmans,1 Klaas Nicolay,1 and Gustav J. Strijkers1,3*

INTRODUCTION

Superparamagnetic iron oxide particles find application as contrast agents for molecular and cellular MRI and are used to facilitate in vivo cell tracking and, among others, inflammation imaging (1–4). The presence of superparamagnetic iron oxide particles in tissue can be visualized by lowered signal intensity or signal voids in T2* or T2*-weighted imaging. This negative contrast poses challenges to the quantitative assessment of iron concentration; moreover, interpretation is conspicuous because signal voids are often difficult to distinguish from image artifacts. Therefore, considerable effort is put in the development of MRI methods that provide a quantitative readout of the presence of iron oxides. Furthermore, quantitative MR imaging enables standardized comparisons of results between different sites and systems. A common approach to quantitative imaging of iron oxides is by mapping the transversal relaxation time T2(*) (5–10).

In this study, we investigated quantitative T1p measurements for imaging of iron oxide-based contrast agents. We hypothesized that T1p provides a more sensitive readout of the presence of iron oxides than T2. T1p is the longitudinal relaxation in the rotating frame of reference during the application of a radiofrequency (RF) pulse parallel to the magnetization vector. T1p-weighted imaging is commonly accomplished using spin-lock MR, for which magnetization is first excited by a 90° pulse and subsequently subjected to a long, continuous-wave B1 spin-lock pulse. Because the B1 field is low compared to the B0 field, the tissue T1p relaxation is mediated by processes with long correlation times such as proton diffusion, proton exchange, and macromolecular tumbling. These processes can be probed by varying the B1 spin-lock field strength; they show up as tissue T1p dispersion as function of B1 (11–16). The T1p relaxation is also influenced by the proton off-resonance frequency (Δω), which introduces a T1 component that depends on the ratio of B1 and Δω (17–20).

From T2 and T2* relaxation models, it is known that iron oxides enhance the proton relaxation rate via their local magnetic susceptibility in a particle size-, coating-, and diffusion-dependent manner (21–23). These effects are categorized into several relaxation regimes. Similar mechanisms will influence T1p relaxation in the presence of iron oxides. We expected that protons in the static regime will experience off-resonance spin lock in the vicinity of a particle. Protons that diffuse through the magnetic field gradients on the timescale of the spin-lock pulse duration, however, were predicted to lose their spin lock because of the fluctuating magnetic field that they experience. We therefore hypothesize that, in the latter diffusion regime, the presence of iron oxides could lead to changes in the tissue T1p dispersion that are larger than T2 changes, potentially providing a more sensitive readout for the detection of iron oxides.
performed using a cryo-holder at 100 K with field emitter gun operated at 300 kV. Cryo-TEM was acquired with a Tecnai F30ST (FEI, Hillsboro, OR) TEM imaging.

particles; MPIO will be referred to as microparticles. Resovist, and ION-Micelles will be referred to as nano-particles; MPIO will be referred to as microparticles.

Sample Preparation

To evaluate this hypothesis, $T_{1p}$ dispersion curves of agarose gels with various concentrations of superparamagnetic iron oxide micro- and nanoparticles were measured and compared to $T_2$ data. Particles with different iron oxide core size, overall particle size, coating thickness, and composition were selected such that different relaxation regimes could be probed (24). Additionally, mice were injected with an iron oxide contrast agent, and quantitative $T_{1p}$ and $T_2$ values of the contrast agent-containing liver were evaluated.

METHODS

Contrast Agents, Characterization, and Properties

Four superparamagnetic iron-oxide contrast agents were chosen for their variation in size and composition: Sinerem (ferumoxtran; Guerbet, Villepoint, France), Resovist (ferucarbotran; Bayer Schering Pharma, Berlin, Germany), iron oxide nanoparticle (ION)-Micelles (home-made (25)), and microparticles of iron oxide (MPIO) (Bangs Laboratories Inc, Fishers, Indiana). Table 1 summarizes relevant characteristics of the particles. Sinerem and Resovist have a similar iron oxide core size (4–6 nm) and coating composition (dextran) but a different hydrodynamic diameter; they were chosen to investigate the effect of the coating thickness on $T_{1p}$ relaxation. ION-Micelles have a much larger core size (25 nm), although their hydrodynamic diameter (61 nm) is comparable to that of Resovist (62 nm), yielding data on the effect of iron oxide core size. The hydrodynamic diameter of MPIO (860 nm) is much larger than the other three particles and it was selected to evaluate $T_{1p}$ dispersion in the static dephasing regime. In this article, Sinerem, Resovist, and ION-Micelles will be referred to as nanoparticles; MPIO will be referred to as microparticles.

Transmission electron microscopy (TEM) images were acquired with a Tecnai F30ST (FEI, Hillsboro, OR) TEM with field emitter gun operated at 300 kV. Cryo-TEM was performed using a cryo-holder at −170 °C. Saturation magnetization values were determined using a vibrating sample magnetometer (ADE Technologies, Newton, MA) at room temperature; sample iron concentrations were obtained with inductively coupled plasma atomic emission spectrometry (ICP-AES).

Sample Preparation

The dilution series for the in vitro nuclear magnetic resonance (NMR) measurements were made in 2% agar gel because of its tissue-like $T_1$ and $T_2$ properties. Agarose (Sigma-Aldrich, St. Louis, MO) was dissolved in demineralized water by heating it to 80 °C, thoroughly mixed with the iron oxide contrast agent solution, and then poured into 5 mm-NMR tubes (Bruker BioSpin, Ettlingen, Germany). During the pouring, the tubes were placed in a water bath at 80 °C to prevent the agar gel from solidifying against the walls of the tubes. Subsequently, the tubes were placed at room temperature to allow the gel to solidify on the bottom of the tube. The iron concentrations (0, 50, 125, 250, and 500 μM) were chosen to be identical for the different contrast agents because it is common practice to assess relaxation enhancement as a function of iron concentration. This procedure consequently results in varying particle concentrations. All samples were made as one single batch, that is, from the same agarose solution.

In Vitro NMR

All experiments were performed with a tabletop NMR spectrometer operating at 60 MHz/1.41T (Minispec60, Bruker BioSpin) with a solenoid RF coil, and the sample temperature was maintained at 20 °C. The $T_{1p}$ sequence consisted of a 90° excitation pulse directly followed by a continuous wave spin-lock pulse and free induction decay acquisition. In order to assess the $T_{1p}$ dispersion, the spin-lock amplitude was varied between 6.5 and 91 kHz. The spin-lock amplitude could not be set directly and was adjusted by changing the RF attenuation. Using a sample of 2% agar gel without contrast agent, the 90° pulse duration was calibrated for each setting of the RF attenuation, and then the spin-lock amplitude $\gamma B_1$ at the set RF attenuation was calculated.

$T_{1p}$ was determined from exponential fitting of peak signal intensities from 10 different spin-lock durations. Spin-lock durations were logarithmically distributed between a minimum of 1 ms and a maximum of 10 to 50 ms, with the maximum depending on $B_1$ and hardware restrictions. For comparison with $T_{1p}$, $T_2$ was acquired using the Carr-Purcell-Meiboom-Gill (CPMG) sequence with 1024 echoes, 4.6 μs 180° pulse duration, and a 90°–180° pulse separation of 40 to 500 μs, resulting in a longest echo time (TE) ranging between 81.92 and 1024 ms. For all in vitro NMR experiments, the repetition time (TR) was set to five times the sample $T_1$, as determined by an inversion recovery experiment.

NMR Data Analysis

The dispersion data were fitted with an exchange model in order to extrapolate the $T_{1p}$ curves to a spin-lock amplitude of 0 kHz for direct comparison with $T_2$ acquired with the CPMG sequence. Relaxation behavior

<table>
<thead>
<tr>
<th>Coating</th>
<th>Sinerem</th>
<th>Resovist</th>
<th>ION-Micelle</th>
<th>MPIO</th>
</tr>
</thead>
<tbody>
<tr>
<td>Iron oxide core diameter (nm)</td>
<td>4–6 (43)</td>
<td>4–6 (29,44)</td>
<td>25 (25)</td>
<td>4–13 a</td>
</tr>
<tr>
<td>Hydrodynamic diameter (nm)</td>
<td>32 (25)</td>
<td>62 (25)</td>
<td>61 (25)</td>
<td>860 b</td>
</tr>
<tr>
<td>Magnetization (Am² kg⁻¹ Fe)</td>
<td>71</td>
<td>89</td>
<td>93</td>
<td>105</td>
</tr>
</tbody>
</table>

aMPIO core diameter determined by TEM analysis.
bMPIO particle diameter provided by manufacturer.
in agar gels can be approximated by a two-pool exchange model with a free water and a bound water pool. The equation describing a two-pool exchange model assuming negligible dipolar relaxation effects and on or near resonance conditions is (11)

\[
R_{1p} = R_1 \cos^2 \theta + \left( R_2^0 + A \frac{r_0}{r_0^2 + \omega_0^2} \right) \sin^2 \theta \tag{1}
\]

with \(R_{1p}\) the longitudinal relaxation rate in the rotating frame \((T_{1p}^{-1})\), \(R_2^0\) the transverse relaxation rate without exchange, \(\theta\) the tilt angle of the effective spin-lock field, \(A = p_1 p_b \Delta \omega_0^2\), the product of the fractions of free and bound water multiplied by the square of their frequency difference, \(r_0\) the exchange rate, and \(\omega_0\) the spin-lock amplitude in rad/s. Assuming on-resonance spin lock \((\theta = \pi/2)\) and substituting \(\frac{1}{\omega_0}\), Equation (1) is rewritten as

\[
R_{1p} = R_2^0 + A \frac{\tau_{ex}}{1 + \omega_0^2 \tau_{ex}^2} \tag{2}
\]

with \(\tau_{ex}\) the exchange time. The dispersion data was fitted to this equation, and \(R_{1p}(\omega_0 = 0)\) was compared to \(R_2\) measured by the CPMG sequence.

To assess the effect on contrast, the normalized change in \(R_{1p}\)

\[
N \Delta R_{1p} = \frac{R_{1pCA} - R_{1p0}}{R_{1p0}} \tag{3}
\]

was determined for each spin-lock amplitude. Here, \(R_{1pCA}\) is \(R_{1p}\) of the sample with contrast agent, and \(R_{1p0}\) is \(R_{1p}\) of the plain agar sample. Similarly, we define the normalized change in \(R_2\)

\[
N \Delta R_2 = \frac{R_{2CA} - R_{2,0}}{R_{2,0}} \tag{4}
\]

for a quantitative comparison of the change in \(R_{1p}\) and \(R_2\) values in the presence of iron oxide contrast materials.

**In Vivo MRI**

Animal experiments were approved by the animal experiment committee of Maastricht University (The Netherlands). Seven female C57Bl/6 mice (Charles River, Maastricht, The Netherlands) with 20.5 to 22.3 g bodyweight (BW) were anesthetized with 1% to 2% isoflurane. A catheter containing 0.9% NaCl with heparin and a 150 μL bolus of 50 μmol Fe/kg BW Sinerem was inserted into the tail vein. Sinerem was chosen for the in vivo experiments because it was one of the three nanoparticles that yielded enhanced contrast in the in vitro experiments. The mice were placed in a supine position on an animal bed with an anesthesia mask and monitored with electrocardiogram (ECG) electrodes attached to the front paws, a balloon respiration sensor, and a rectal temperature probe. The body temperature was maintained at approximately 37°C with a warm water pad.

In vivo MRI experiments were performed with a 9.4T small animal scanner equipped with a 35-mm-diameter quadrature birdcage RF coil (Bruker BioSpin). An axial slice (field of view \(= 22 \times 22 \text{ mm}^2\), matrix \(= 220 \times 220\), thickness \(= 1 \text{ mm}\) was positioned just under the diaphragm to encompass a large section of the liver. The liver was chosen because Sinerem is excreted via the hepatobiliary pathway and will thus accumulate in the liver and spleen. All acquisitions were respiratory-gated and ECG-triggered, with the acquisition window positioned in the end-diastolic phase of the cardiac cycle in order to minimize motion artifacts caused by cardiac and respiratory motion. A fast low-angle shot (FLASH) \((TR = 1 \text{ cardiac cycle} \ (95 - 135 \text{ ms}), TE = 3.2 \text{ ms}, \ \text{flip angle} \ [FA] = 40^\circ, \ \text{and number of averages} \ [NA] = 3)\) image was acquired to serve as anatomical reference.

The \(T_{1p}\) sequence consisted of a \(B_1\) and \(B_0\) field inhomogeneity-compensating spin-lock preparation (26), followed by a fast imaging with steady state precession (FISP) readout. The imaging acquisition window of approximately 11 ms, preceded by the global spin-lock preparation, was kept at a constant position in the cardiac cycle to prevent motion artifacts and guarantee accurate registration between images recorded with different spin-lock times. The FISP readout was performed in 22 segments of five echoes. Other sequence parameters were: \(TE = 1.33 \text{ ms}, \ TR = 2.67 \text{ ms}, \ \text{segment TR} = 2000 \text{ ms}, \ NA = 3, \ \text{acquisition matrix} = 110 \times 110, \ \text{reconstruction matrix} = 220 \times 220, \ \text{FA} = 30^\circ. \ \text{The acquisitions were performed for five different spin-lock amplitudes} \ ((\gamma B_1 = 100, 250, 500, 1000, \ \text{and} \ 1500 \text{ Hz}) \ \text{and with five different spin-lock times} \ [(\text{TSL}) = 3, 6, 12, 24, \ \text{and} \ 48 \text{ ms})]. \ \text{Additionally, an experiment with a spin-lock amplitude of} 0 \text{ Hz was performed to serve as a} T_2 \text{ measurement because under this condition it turns into a spin-echo preparation, which is} T_2\)-weighted.

After acquisition of the last preinjection image, the contrast agent bolus was injected manually. Subsequently, a second FLASH image was acquired to confirm successful intravascular injection, after which the mouse was removed from the scanner and allowed to recover. Twenty-four hours later, the mouse was anesthetized again and postinjection images were acquired. While still under anesthesia, the mice were euthanized directly after the last MRI scan.

**MRI Data Analysis**

Image analysis was performed in Mathematica 9.0 (Wolfram Research, Champaign, IL). Pixel-wise mono-exponential fitting of the signal intensities at different TSL was performed for each spin-lock amplitude to generate \(T_{1p}\) and \(T_2\)-maps. Regions of interest (ROIs) were drawn manually around the liver and muscles on the FLASH anatomical reference images and copied to the \(T_{1p}\) and \(T_2\)-maps to determine the mean liver and muscle \(T_{1p}\) and \(T_2\), respectively. The main blood vessels of the liver and also the gall bladder, when visible in the slice, were excluded from the liver ROI. Pixels with an \(R^2\) of fit \(<0.7\) were excluded from further analysis. Histograms of \(T_{1p}\) and \(T_2\) in the liver were generated for each mouse with the following settings: 20 bins, bin width 5 ms, and a range of 0 to 100 ms. A spline function was fitted through the averaged histograms to visualize the changes in \(T_{1p}\) and \(T_2\) distributions that were induced by Sinerem injection.

**Statistics**

Statistical analysis was performed in SPSS Statistics 22 (IBM, Armonk, NY). In vivo data on pre- and
postinjection T1<sub>p</sub> and T2 values were compared with a two-sided t test for each spin-lock amplitude. The effect of spin-lock amplitude on both the pre- and postinjection T1<sub>p</sub> datasets was tested for significance using an analysis of variance (ANOVA) for repeated measures with the spin-lock amplitude as the within-subject factor. In case a significant effect of spin-lock amplitude was found with ANOVA, the individual T1<sub>p</sub> levels were compared with a one-sided paired t test. The level of significance was set to α = 0.05 for all tests.

RESULTS

Figure 1 shows the cryo-TEM images of the three types of nanoparticle. The iron oxide cores (black dots and discs), but not the coatings, are visible in the images. Sinerem seemed evenly dispersed with little or no aggregation of cores (Fig. 1A), which indicates that most particles contain a single core. For Resovist, the dispersion is also relatively uniform (Fig. 1B). However, alongside single cores, aggregates of two to three cores—and occasionally even larger aggregates—were observed for Resovist. This suggests that a subset of Resovist particles contain more than one iron oxide core. ION-Micelles also displayed a few small aggregates (Fig. 1C). However, the limited spacing between clustered cores indicated an aggregation of coated particles rather than single particles containing multiple cores. Higher magnification TEM images in Figure 2 show the core-shell structure of Sinerem and MPIO. Sinerem consists of spherical particles with a single core of iron oxide (Fig. 2A). MPIO are polymer beads with a broad size distribution that contain multiple iron oxide cores per particle, which can be well distinguished in the inset of Figure 2B. The saturation magnetization values for each of the contrast agents can be found in Table 1.

In Figure 3, T1<sub>p</sub> dispersion curves are shown for the in vitro experiments with different types and concentrations of iron oxides. Agar without contrast agent displayed the strongest dispersion. Over the full range of spin-lock amplitude γB1, T1<sub>p</sub> increased from 91.7 ms (at 6.5 kHz) to 479.4 ms (at 91 kHz). The data was fitted to the exchange model for dispersion in agar gels described by Equation (2). The average exchange rate 1/τ<sub>ex</sub> observed in plain agar was 143 ± 1 kHz, which is in the range of values reported for the dominant exchanging water class in agar gels (27).

For all contrast agents, T1<sub>p</sub> decreased with increasing iron concentration. T1<sub>p</sub> at the highest iron concentration (500 μM) and maximum spin-lock amplitude (91 kHz) was 24.2 ms for Sinerem (Fig. 3A), 15.6 ms for Resovist (Fig. 3B), 4.1 ms for ION-Micelle (Fig. 3C), and 233.2 ms for MPIO (Fig. 3D). Furthermore, in the presence of Sinerem, Resovist, and ION-Micelles, the T1<sub>p</sub> dispersion curves were notably flatter. Solid lines in Figure 3 are exchange model fits using Equation (2). Extrapolated values for T2 = T1<sub>p</sub>(0 kHz) agreed with the T2 values as measured with the CPMG sequence (Fig. 3A–C), with the exception of MPIO, for which T1<sub>p</sub>(0 kHz) was higher than CPMG T2 (Fig. 3D).

The strong reduction in T1<sub>p</sub> dispersion at higher iron oxide concentrations resulted in strong T1<sub>p</sub> differences compared to 0 μM iron oxide at the higher spin-lock amplitudes. In other words, a higher T1<sub>p</sub> contrast is generated by the iron oxides for the higher spin-lock amplitudes. To quantify this contrast, the normalized difference in the T1<sub>p</sub> relaxation rate (NΔR<sub>1p</sub>, Eq. [3]) as function of iron concentration for the different iron oxides and varying spin-lock amplitude is shown in Figure 4. NΔR<sub>1p</sub> was highly linear with iron concentration (R² > 0.96). With the exception of MPIO, all contrast agents yielded enhanced NΔR<sub>1p</sub> for increased spin-lock amplitude. NΔR<sub>1p</sub> was enhanced up to 5.1-fold (Sinerem), 5.9-fold (Resovist), and 4.7-fold (ION-Micelle) compared to NΔR<sub>2</sub>. Notably, for the three nanoparticles, NΔR<sub>1p</sub> was larger than NΔR<sub>2</sub>—even for the smallest spin-lock amplitude. For MPIO, the absolute T1<sub>p</sub> decrease was small (Fig. 3D), and a significant decrease in dispersion was not observed. This was also reflected in the very low NΔR<sub>1p</sub> and NΔR<sub>2</sub> values (Fig. 4D), indicating that only minor changes in T1<sub>p</sub> and T2 contrast were achieved by the addition of microparticles of iron oxide.

Figures 5A,C show cross-sectional in vivo liver T1<sub>p</sub>-weighted images of mice pre- and 24 h postinjection of
Sinerem. In Figures 5B,D, corresponding T1r maps for a spin-lock amplitude of 500 Hz are presented. Twenty-four hours after contrast agent injection, T1r-weighted signal intensity in the liver had markedly dropped, and quantitative T1r values were substantially lower. Quantitative T1r values from ROIs in the liver and muscle as a function of spin-lock amplitude are displayed in Figures 6A,B, respectively. Both pre- and postinjection mean liver T1r displayed a dispersion, with increasing T1r (P < 0.001) toward higher spin lock. Differences between pre- and postinjection T1r values were significant (P < 0.05) for all spin-lock amplitudes except for 100 Hz (P = 0.194) and 1500 Hz (P = 0.059). Mean muscle T1r showed no significant dispersion with spin-lock amplitude (P = 0.357), and no significant differences were found between pre- and postinjection time points. All T2 values were significantly lower than the T1r values; and pre- and post-T2 values were significantly different in the liver. Figures 6C,D show typical T1r and T2 signal decay curves in liver and muscle, respectively. Apparently independent from T1r and T2 decay, large differences in pre- and postinjection signal intensities were observed for the liver (Fig. 6C). In muscle T1r and T2 values were not influenced by injection of Sinerem. In Figure 7, average T1r and T2 distributions in the liver are shown. At 24 hours after injection, a small but significant shift in the histograms toward lower relaxation values was observed for T1r at all spin-lock amplitudes and T2.

**DISCUSSION**

**In Vitro Experiments**

In this study, we investigated the effect of four iron oxide contrast agents on T1r. In vitro experiments in agar phantoms demonstrated both a decrease in T1r values and a loss of T1r dispersion in the presence of iron oxide nanoparticles. For MPIO, no significant changes in T1r dispersion were observed. Due to the loss of dispersion, a higher contrast was generated at high spin-lock amplitudes. The normalized contrast (NDR1r) proved linear with iron concentration. For all three nanoparticle contrast agents, NDR1r was higher than NDR2 and was found to increase with spin-lock amplitude. The fact that the extrapolated values for T2 = T1r(0 Hz) matched the T2 values measured by CPMG is in accordance with T1r relaxation theory (15).

**Iron Oxide T1p Contrast Mechanism**

We believe that the mechanism underlying enhanced iron oxide T1p contrast is loss of spin lock, leading to the reduction of native tissue T1p dispersion. The effect of iron oxides on T2(*) relaxation has been extensively studied, and the effects of particle size and coating composition are well understood (21–24). Important factors for the R2* and R2 relaxation rates are the strength and spatial extent of local field gradients, proton diffusion during the time of the NMR experiment, and the relative proportion of these factors. Several relaxation regimes can be distinguished, but the discussion here will focus on those regimes relevant for the present experiments. For a full discussion of all R2(*) regimes, we refer to refs. (22) and (23).

The first regime that is of importance for the iron oxide T1p contrast mechanism is the static dephasing regime (SDR). In this regime, which is generally applicable for large particles, proton displacement by diffusion is small relative to the spatial scale of field gradients. Hence, protons near the particle essentially experience a static field offset. For R2* relaxation, this means rapid dephasing and R2* approaches a maximum. R2 is low because the static field offset can be effectively compensated through a refocusing pulse. In the case of R1p during spin-locking, a static field offset will cause off-resonance spin lock along an effective magnetic field Beff, with relaxation along this field (17). This off-resonance relaxation rate R1p/off becomes a combination of R1 and R1p, in a ratio dependent on the ratio of B1 and...
FIG. 3. In vitro $T_{1p}$ dispersion curves. $T_{1p}$ as function of $\gamma B_1$ for different concentrations of (A) Sinerem, (B) Resovist, (C) ION-Micelle, and (D) MPIO in 2% agar gel. Solid lines are exchange model fits using Eq. [2], which represents the native $T_{1p}$ dispersion of agar. Both $T_{1p}$ values and $T_{1p}$ dispersion decreased with increasing iron concentration. The data points located to the left of the horizontal axis break (solid symbols) are $T_2$ values measured by CPMG.

FIG. 4. $T_{1p}$ contrast. $N_D R_1$ as a function of iron concentration $[Fe]$ for varying spin-lock amplitudes for (A) Sinerem, (B) Resovist, (C) ION-Micelle, and (D) MPIO. The solid symbols are $N_D R_2$. Solid lines are linear fits of the data ($R^2 > 0.96$). For all iron oxides, $N_D R_1$ increased linearly with iron concentration. $N_D R_1$ was higher than $N_D R_2$ for all preparations, with the exception of MPIO. Note that the vertical axes have different scaling.
Off-resonance spin lock is often applied to decrease specific absorption rate (SAR) because a similar $B_{\text{eff}}$ can be reached with a lower $B_1$ (19). In this case, however, the source of off-resonance is the contrast agent and not the RF pulse. Because $B_1$ remains unchanged, a higher $B_{\text{eff}}$ is achieved in the vicinity of the particle. The field offset is static; therefore, no loss of spin lock and thus no effect on the native $T_{1p}$ dispersion of a tissue are to be expected in the SDR.

The other regime that is considered important for the experiments in this study is the so-called visit-limited regime (VLR); it marks the transition between the SDR and the motional averaging regime, which will not be discussed here (22). Iron oxide nanoparticles in the VLR are surrounded by a full dephasing zone in which the proton spins are fully dephased upon entry. According to our hypothesis, the gradients and diffusional motion will lead to a loss of spin lock in the full dephasing zone. Experimental observations (Figs. 3 and 4) are in agreement with this hypothesis. First, strong dephasing in the full dephasing zone in the presence of the iron oxide nanoparticles leads to a rapid decrease in signal intensity as a function of TSL, resulting in a lower measured $T_{1p}$ value. Furthermore, because of the loss of spin lock during TSL, signal decay and thus $T_{1p}$ become essentially independent from the spin-lock amplitude, resulting in a loss of the exchange-mediated dispersion with $\gamma B_1$ of native agar. Moreover, $N_\Delta R_{1p}$ was linear with concentration, which can be explained by a linear increase of the total volume occupied by the full dephasing zones surrounding the particles with increasing concentration.

$\Delta \omega$ (20). Off-resonance spin lock is often applied to decrease specific absorption rate (SAR) because a similar $B_{\text{eff}}$ can be reached with a lower $B_1$ (19). In this case,
The VLR is applicable to particles at the transition between the motional averaging regime and the SDR. This transition is defined by

$$\frac{r_p^2}{D} \approx \frac{\pi \sqrt{15}}{4 \Delta \omega_z}$$  \[5\]

with $r_p$ the radius of a nanoparticle in the VLR, $D$ the diffusion coefficient, and $\Delta \omega_z$ the root-mean-squared frequency shift at the particle surface (28).

The diffusion coefficient in a 2% agar gel is of the same order of magnitude as that of free water ($\sim 2.5 \times 10^{-9} \text{ m}^2/\text{s}$), and the frequency shift at the particle surface caused by its magnetization is on the order of that of magnetite ($3.0 \times 10^7 \text{ rad/s}$). This means that relaxation of water in the proximity of particles with a diameter of around 30 nm occurs in the visited limiting regime (22, 28). With its much larger diameter of 860 nm, MPIO is therefore in the SDR. With hydrodynamic diameters between approximately 30 and 60 nm (Table 1), the VLR applies to Sinerem, Resovist, and ION-Micelles (23).

**Comparison of the Four Contrast Agents**

Comparing different nanoparticles, $N_1 R_{1p}$ was higher for Resovist than for Sinerem (Fig. 4). Differences in the observed $T_{1p}$ contrast between the different iron oxide formulations can be related to their different physico-chemical properties (Table 1). Sinerem and Resovist have comparable iron oxide core sizes, and both have similar dextran-based coatings. However, the coating of Resovist is twice as thick as the one of Sinerem.
Therefore, the full dephasing zone of Resovist may occupy a larger volume, leading to a stronger $T_{1p}$ decrease and more pronounced loss of dispersion as compared to Sinerem (23). Also, the saturation magnetization for Resovist is higher (89 vs. 71 Am$^2$ kg$^{-1}$ Fe). Additionally, Resovist has a broader size distribution than Sinerem and contains a small fraction (~3%) of particles with a larger iron core up to 30 nm (29,30). Also, multiple cores per particle (Fig. 1B) are found, leading to an effectively higher magnetic moment per particle, and therefore stronger $T_{1p}$ decrease with concentration.

ION-Micelles displayed the highest $N\Delta R_{1p}$ of the four iron oxide formulations. The main reason can be found in the larger size of the particle's core and the higher magnetic moment per particle. The monocristalline core of ION-Micelles is approximately five times larger in diameter (25 nm) than those of Sinerem and Resovist (both 4–6 nm). A larger core combined with a higher saturation magnetization (93 Am$^2$ kg$^{-1}$ Fe) results in a much larger magnetic moment per particle, with larger local field gradients and a larger affected volume for each individual particle. The amphiphilic phospholipid coating with hydrophilic polyethylene-glycol (PEG) tails constitute a region of restricted diffusion similar to the dextran for Sinerem and Resovist (31). However, this coating is thinner than the dextran coatings; therefore, the main reason for the larger $N\Delta R_{1p}$ is probably the much larger magnetic moment of ION-Micelles.

For MPIO, a minor $N\Delta R_{1p}$ and no significant change in $T_{1p}$ dispersion were observed. MPIO consists of multiple small iron oxide cores incorporated in a polymer mesh with an outer layer of pure polymer (Fig. 2B). The particles can be regarded as having one very large (860 μm) superparamagnetic core (28), producing an equally large magnetic moment. The polymer mesh is not water permeable; thus, water protons are restricted to the outer regions of the gradient field where the magnetic moment is lower and there is less spatial fluctuation than close to the center. These particles are in the static-dephasing regime; rather than inducing a loss of spin lock, they will preserve off-resonance spin lock and $T_{1p}$ dispersion. Also, because of their lower surface-area-to-volume ratio (circa 0.007 vs. 0.1–0.2 for the nanoparticles), a much smaller effective volume, and thus a smaller portion of protons in the sample, are affected. Additionally, because the iron concentration was kept constant for different contrast agents and the MPIOs have a larger iron load, the particle concentration was much lower, which further adds to a smaller $T_{1p}$ change.

In Vivo Iron Oxide $T_{1p}$ Contrast

The in vivo experiments revealed minor $T_{1p}$ dispersion in the mouse liver both before and 24 hours after intravenous injection of a bolus of Sinerem (Fig. 6A). The $T_{1p}$ distribution in the liver shifted toward lower values after injection for all spin-lock amplitudes (Fig. 7), yielding a significantly decreased mean liver $T_{1p}$ at spin-lock amplitudes of 250, 500, and 1000 Hz. These changes of in vivo $T_{1p}$ values were only small and were not significantly different from the reduction of in vivo $T_2$ values.

The in vitro advantage of $T_{1p}$ over $T_2$ was thus not observed in vivo for Sinerem accumulation in the liver.

In vivo $T_{1p}$ dispersion in the liver did not appear to be influenced by the presence of Sinerem in a similar fashion as for the agar experiments. One explanation for this may be found in the in vivo fate of the particles. Sinerem is mainly taken up by Kupffer cells in the liver, resulting in a clustered distribution rather than the even dispersion in agar gels in vitro (32). Similar to the clustering of iron oxide cores inside the MPIO particles, these clusters might behave as larger particles with a SDR relaxation mechanism (28). Indeed, it is known that such in vivo clustering results in reduced changes in $R_1$ and $R_2$ values (32,33). Another cause may be found in the low precontrast dispersion, which makes it difficult to detect significant dispersion changes. Additionally, due to hardware restrictions, only a low spin-lock amplitude (1500 Hz) could be applied in vivo, limiting the range over which the dispersion could be studied.

The liver contrast between pre- and postinjection in the $T_{1p}$-weighted images appeared larger than what could be expected on the basis of quantitative $T_{1p}$ differences (Fig. 5). The reason is a $T_2^*$ effect in the FISP imaging readout, which leads to a shift of the whole $T_{1p}$ and $T_2$ relaxation curves toward lower signal intensity values (Fig. 6C), independent of changes in the quantitative $T_{1p}$ and $T_2$ values.

The fact that no loss of $T_{1p}$ dispersion was observed in the liver does not preclude in vivo application of $T_{1p}$ imaging of iron oxide contrast agents in other tissues such as articular cartilage (34,35), myocardial tissue (36), breast tissue (37), and certain tumors (38) that display larger endogenous $T_{1p}$ dispersions. Furthermore, techniques such as adiabatic $T_{1p}$ and relaxation along a fictitious field (35,39–41) may enable higher effective spin-lock amplitudes to allow probing of higher frequency dispersion regimes without violating hardware restrictions. These techniques also have the advantage that they lower the SAR, which hampers clinical application of conventional $T_{1p}$ imaging. However, the effects of iron oxides on $T_{1p}$ behavior using these sequences remain to be investigated.

CONCLUSION

In the in vitro experiments, it was proven that iron oxide nanoparticles cause loss of spin lock, resulting in the suppression of $T_{1p}$ dispersion. Thereby, $T_{1p}$ contrast is enhanced compared to $T_2$. Spin-lock MR is therefore a promising technique for the sensitive detection of iron oxide contrast agents. However, evaluation in the mouse liver did not reveal an improvement in $T_{1p}$ sensitivity above $T_2$ for iron oxide in vivo, which was probably due to a lack of initial $T_{1p}$ dispersion in the liver and limitations in available spin-lock power.

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